

## UNIT-II

### Magnetic susceptibility:

(Magnetic susceptibility is a dimensionless proportionality constant <sup>The degree of magnetization</sup> of a material in response to applied magnetic field). (It caused by interactions of electrons and nuclei with the externally applied magnetic field)

### Definition:

\* In electromagnetism magnetic susceptibility is defined as

\* The measure of how much a material will be magnetized in an applied magnetic field. It is denoted by  $\chi$ .

$\chi$  <sub>magnetic susceptibility</sub>

### Formula:

The mathematical definition of magnetic susceptibility is the ratio of magnetization to applied magnetizing field intensity. This is a dimensionless quantity.

$$\chi = M/H$$

where,

$\chi$  = magnetic susceptibility.

$M$  = magnetization

$H$  = field intensity.

### Magnetic moment:

Magnetic moment also known as magnetic dipole moment is the measure of (the object's tendency to align with a magnetic field).

Magnetic moment is defined as magnetic strength and orientation of a magnet or other object that produces a magnetic field.

The magnetic moment is a vector quantity. The objects have a tendency to place themselves in such a way that the magnetic moment vector becomes parallel to the magnetic field lines. The direction of the magnetic moment points from the south to the north pole of a magnet. The magnetic field created by a magnet is directly proportional to the magnetic moment.

### Method to produce magnetic moment

Magnetic moment is produced by the following two methods.

- \* The motion of electric charge
- \* Spin angular momentum.

### Measurement of magnetic moment:

Magnetic moments are typically measured by instruments known as magnetometers. (But not all magnetometers are aligned to measure the magnetic moment directly) (some of these devices measure magnetic fields only from the measured magnetic field, the magnetic moment is measured).

### Formula:

The magnetic moment is a vector relating torque of an object to the magnetic field. This is mathematically represented as:

$$\boxed{\tau = M \times B}$$

where  $\tau$  = The torque acting on the dipole

$M$  = magnetic moment

$B$  = external magnetic field.



Unit:

In the definition for the current loop (the magnetic moment is the product of the current flowing and the area  $M = IA$ )

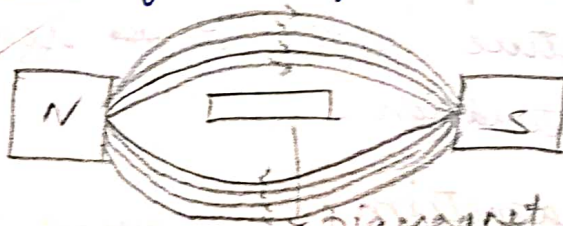
\* So the unit conferring to this definition is articulated by Ampm<sup>2</sup>

\* It can also be suggested in terms of torque and moment conferring to that the torque is measured in joules (J) and the magnetic field is measured in Tesla (T) and thus the unit is JT<sup>-1</sup>

\* So these two units are equivalent to each other and are provided by (Ampm<sup>2</sup> = JT<sup>-1</sup>)

### Diamagnetism:

Diamagnetism substances have a tendency to move from stronger part to the weaker part of the external magnetic field. We can also say that the diamagnetic substance get repelled by a magnet



Consider the figure shown above. we have a diamagnetic substance placed in an external magnetic field. we see that the field lines get ~~is~~ expelled by the material is reduced. If we place this substance in a non-uniform magnetic field. it tends to move from the point of a high electric to that of low electric field.

Magnetic  
Moment  
Ampm<sup>2</sup>  
JT<sup>-1</sup>

## para magnetism:

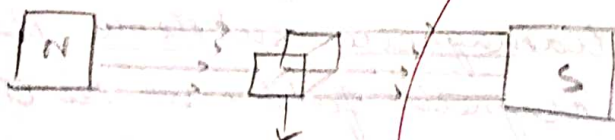
para magnetic substance are those substances that get weakly magnetized in the presence of an external magnetic field.

In the presence of an external magnetic field these substances tend to move from a region of a weak to a strong magnet field.

In other terms, we can say that these substance tend to get weakly attracted to a paramagnetic magnet. In a paramagnetic material, the individual atoms possess a dipole moment which when placed in a magnetic field interact with one another and get spontaneously aligned in a common direction, which results in its

magnetism. As per the Curie law, the magnetism of a paramagnetic substance is inversely proportional to the absolute temperature until it reaches a state of saturation.

## Ferrromagnetism:



ferromagnetic specimen

Ferrromagnetic substance are those substance that when it placed in an external magnetic field get strongly magnetized.

Also they tend to move from a region of



Strong magnetic ~~field~~ <sup>force</sup> to the strong magnetic field and get strongly attracted to a magnetic field the atoms interact with one another and get spontaneously aligned in a common direction. The direction is common even a macroscopic volume which we term as a domain. The domain has a net magnetization and each domain directs itself, which results in its strong magnetization.

### Applications of magnetic properties in solving structural problems:

\* The magnetic property of a material is the atomic (or) subatomic response of a material to an applied magnetic field. where the electron spin and charge create a dipole moment and a magnetic field.

\* Sometimes magnetic properties constitute supplementary information to support other data such as ~~micro~~ <sup>micro</sup> analysis.

#### i) metalloenzymes:

nearly half the important enzymes contain one or more transition metals at the active site. Because of the large molecular weight it is frequently impossible to determine how many metal atoms there are from simple chemical analysis.

\* Knowledge of number of ion, atoms present is important in understanding the function and mechanism of enzymes. If there were two iron atoms the magnetism would be twice as great than if only one iron were present.

\* The enzyme urease was found to contain a dimeric nickel(II) centre based only on its magnetic properties in large proteins such as ferritin strong coupling is seen in a -O-phosphate center, rather like a small piece of rust enclosed in a protein sheath.

\* Individual metal atoms reveal their presence and bonding environment in their <sup>magnetic</sup> ~~field~~ dependent magnetic properties deviation from Curie law.

\* The magnetic susceptibility of copper(II) acetate deviates dramatically from the Curie law. It was analysed as a binuclear rather than monomeric structure long before, X-ray crystallography confirmed this.

Determination of magnetic susceptibility  
by Quoy's method:

The application of a magnetic field to a (paramagnetic substance causes magnetic dipoles to be lined up in the field) directions with (diamagnetic material



the effects is one of polarization of the electron charge cloud) In both cases a magnetic flux density. B is set up and is given by

$$B = \mu H \quad \text{where,}$$

$\mu$  = permeability of the medium and

$H$  = Applied field strength.

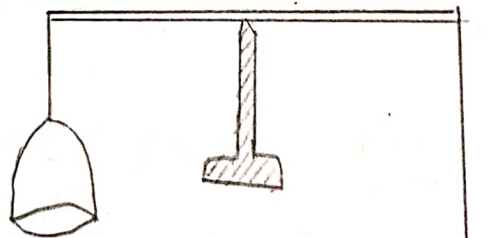
The magnetic susceptibility ( $\chi$ ) is defined as

$$\chi = \mu_r - 1$$

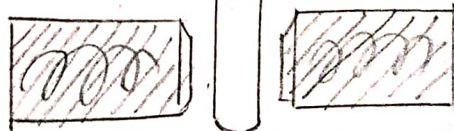
$$\chi = \mu_r - 1 \equiv M/H$$

where  $\mu_r$  = relative permeability of material

The Gouy's method measures the effects of an inhomogeneous magnetic field upon the specimen. (which is suspended from the arm of a sensitive balance by a fine silver chain) The specimen is in the form of a rod of uniform cross sectional area ( $A$ ) (where each end of the rod is in a region of uniform magnet ( $H_0$ ) and that at the centre of the magnetic field ( $H$ ) The magnetic susceptibilities of the specimen is obtained from the relationship.



$$F = \frac{1}{2} \mu_0 A \chi (H^2 - H_0^2)$$



Gouy apparatus for the measurement of magnetic susceptibility.

The specimen, which may be crystalline or liquid is contained in a flat bottomed Gouy tube constructed of glass about 10-15 cm in length and of uniform cross sectional diameter (3-10 mm) The tube itself being a hollow. specimen experiences a force when the sample is weighed in the field. The measurements are also made in air, which has an appreciable susceptibility amount must be taken of the volume of the air displaced on taking these into consideration we get.

$$F = \frac{1}{2} \mu_0 A (\chi_0 - \chi_a) (H^2 - tb^2) + \delta$$

where,  $\chi_0$  is the magnetic susceptibility of air  
 $\delta$  is the tube correction factor.